



U.S. DEPARTMENT OF  
**ENERGY**

Office of  
Science

# Basic Research Needs for Catalysis Science to Transform Energy Technologies

Introduction/Background

Highlights from May 2017 Workshop

Priority Research Directions

Progress on Final Report

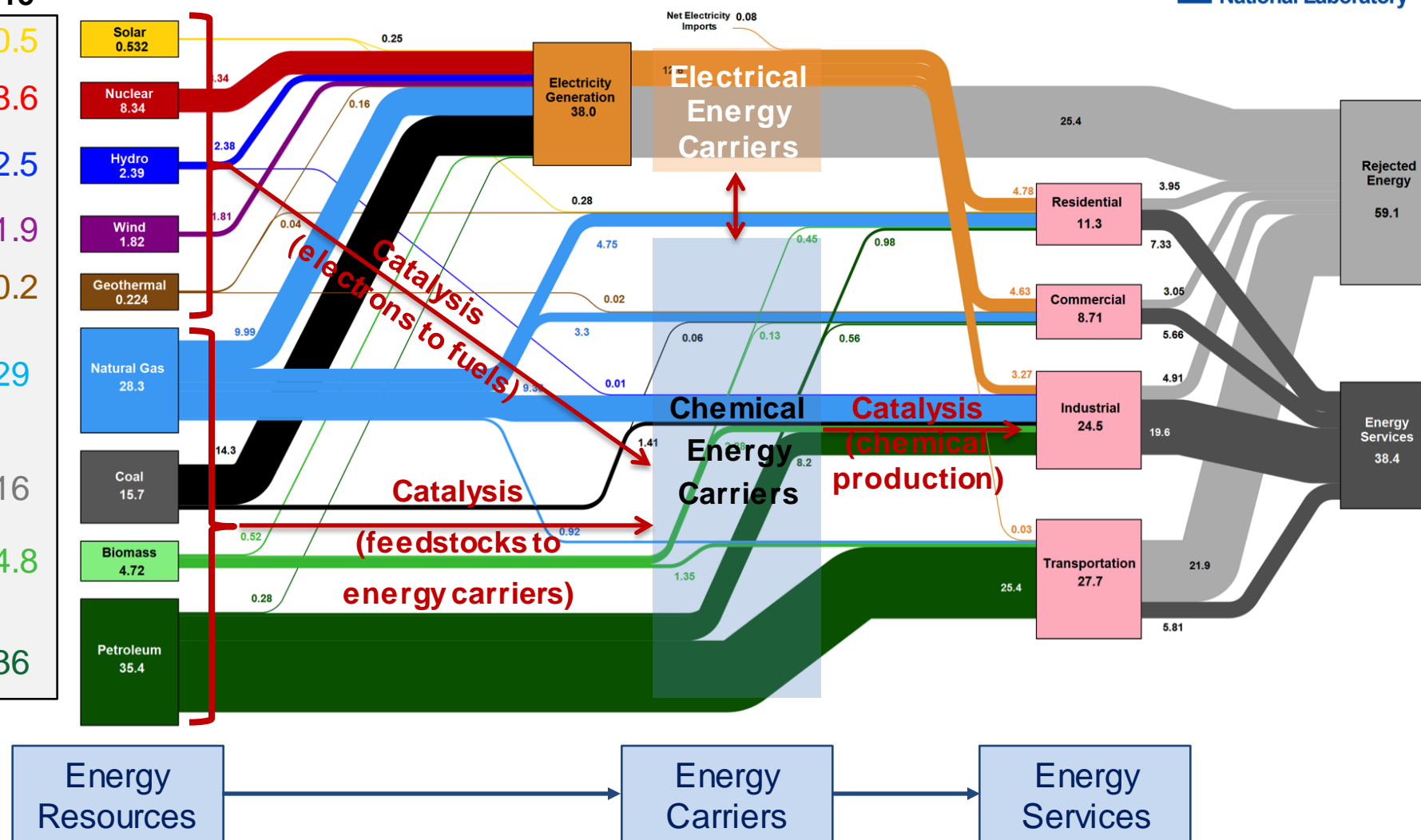
**Prof. Carl A. Koval – University of Colorado at Boulder**

# Changes in Our Energy Landscape Drive the Need for Advances in Catalysis

**% Contribution**  
2007 2015

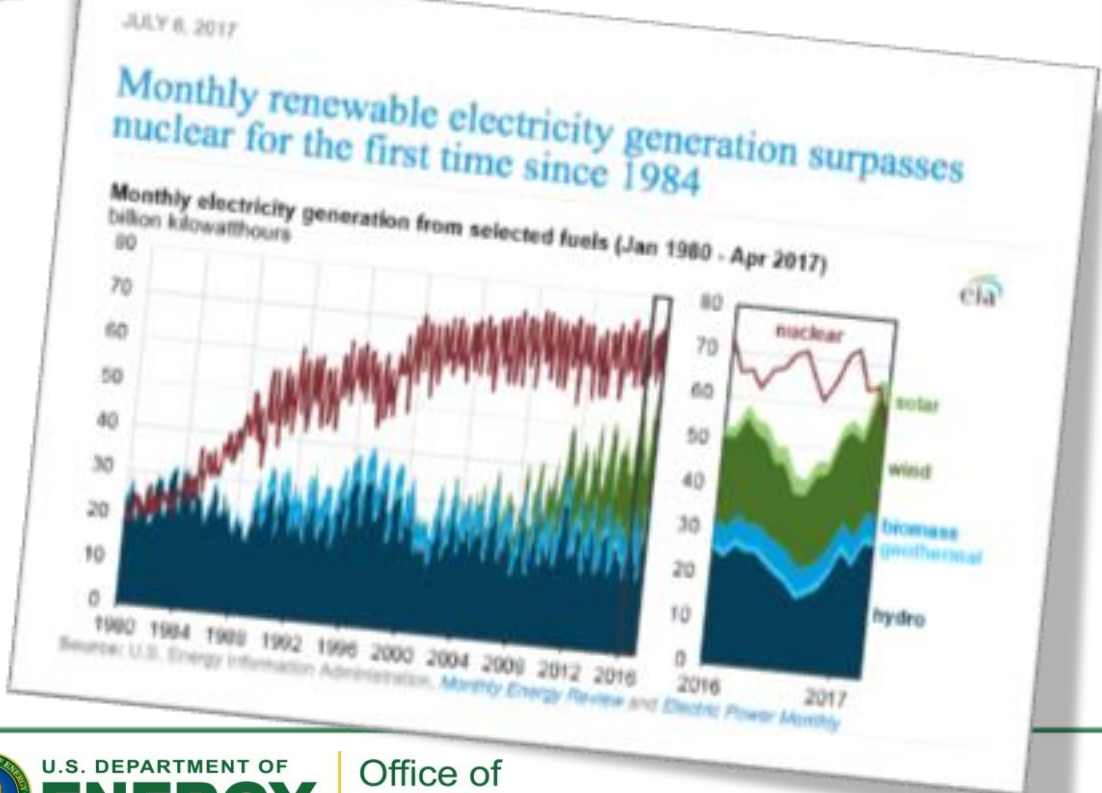


Estimated U.S. Energy Consumption in 2015: 97.5 Quads



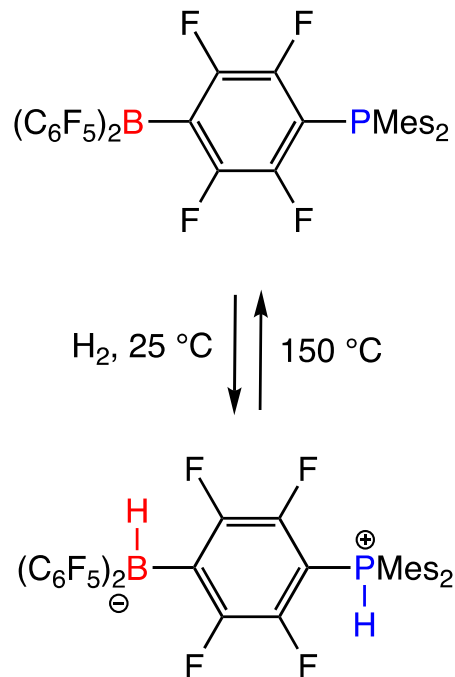
# Impact of New Energy Technologies on the US Chemical Industry

**U.S. Shale Just Triggered A Chemical Industry Renaissance**  
 By Tsvetana Paraskova - Jul 03, 2017, 5:00 PM CDT  
 oilprice.com



# New Paradigms in Catalysis Science

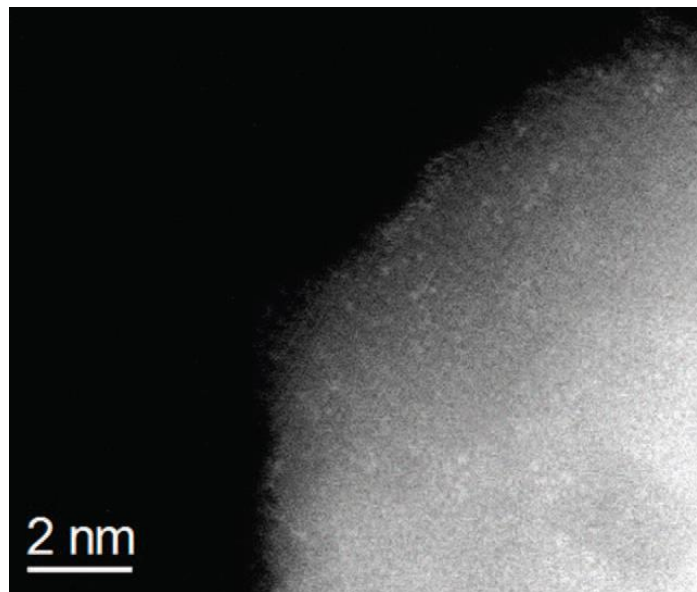
## Metal-free redox catalysis using Frustrated Lewis Pairs (FLPs)



Room temperature  $\text{H}_2$  activation leads to hydrogenation of unsaturated organic compounds.

Stephan, *Acc. Chem. Res.* **2015**, *48*, 306-316.

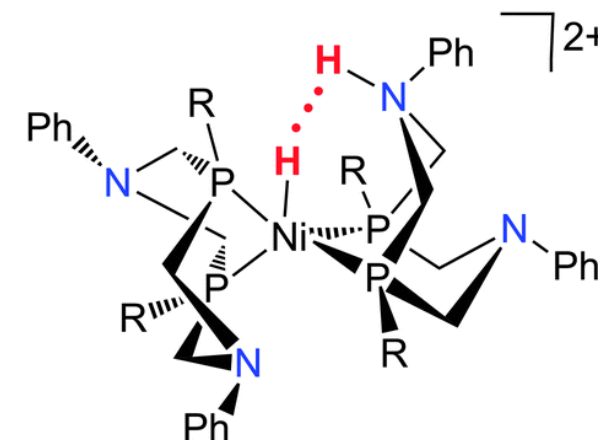
## Stabilization of isolated metal atoms on oxide supports



Individual Pt atoms (bright dots) are stabilized on the surface of  $\text{CeO}_2$  and catalyze CO oxidation without sintering.

Datye et al., *Science* **2016**, *353*, 150-154.

## Outer coordination sphere assistance in multiproton, multielectron reactions



The rate of electrochemical  $\text{H}_2$  evolution is dramatically enhanced by strategic placement of pendant amines that function as proton relays to a metal hydride.

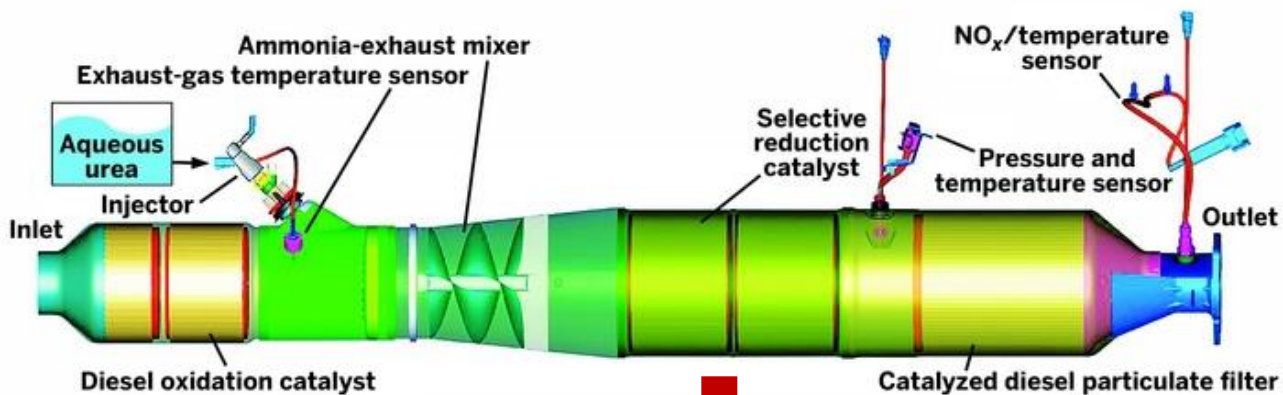
Bullock et al., *Chem. Commun.* **2014**, *50*, 3125-3143.





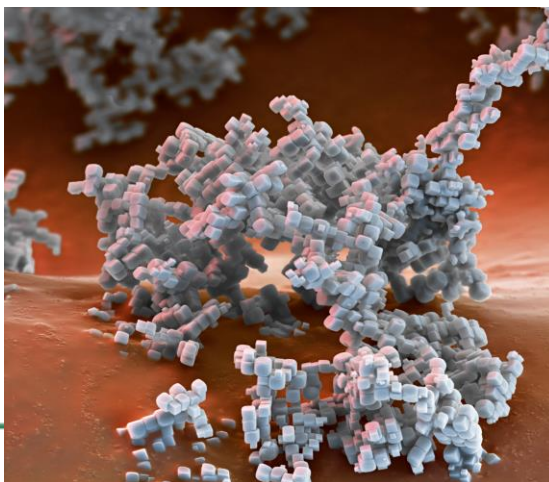
# Low Temperature Catalysts for Diesel Exhaust Aftertreatment

Fuel-efficient diesel vehicles created new challenges for catalysis. Starting in 2007, NO<sub>x</sub> had to be removed from exhaust gases to meet stricter air quality standards. In selective catalytic reduction (SCR), a Cu/zeolite catalyst uses NH<sub>3</sub> to reduce NO<sub>x</sub> to N<sub>2</sub>.



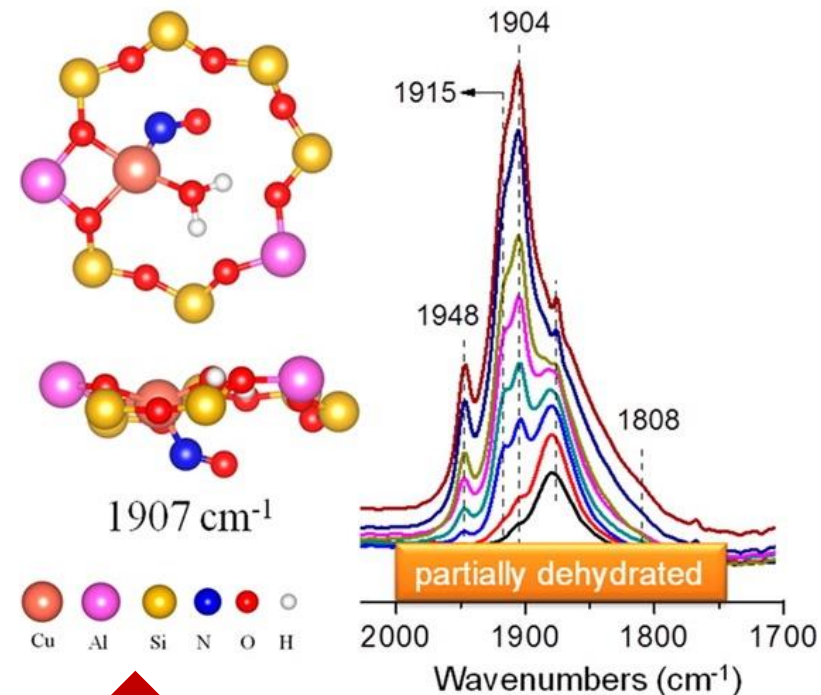
Ford DOC-SCR-DPF system layout  
Beale et al., *Chem. Soc. Rev.* **2015**, *44*, 7371

*Bifunctional Cu/chabazite catalyst, selective for SCR and resistant to poisoning due to Cu<sup>2+</sup> confinement in small zeolite pores*



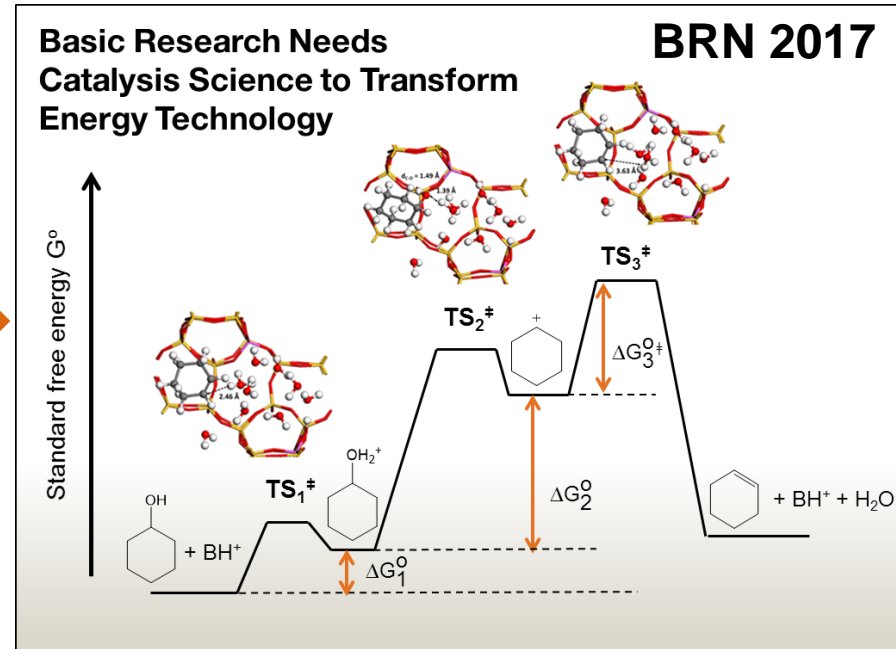
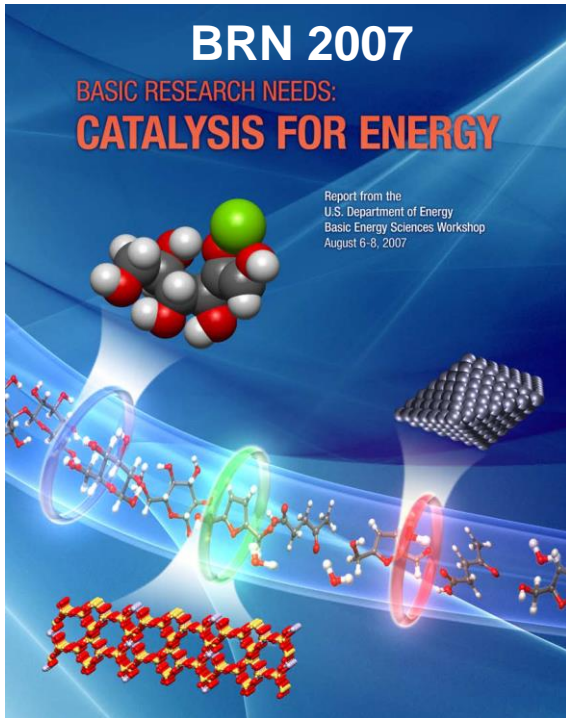
www.basf.com

McEwen, Peden et al., *ACS Catal.* **2014**, *4*, 4093



# New Catalysis Science to Transform Energy Technologies

From feedstock focus to understanding and using chemical complexity.



**Carl Koval**

University of Colorado, Boulder



**Susannah Scott**  
UC Santa Barbara



**Johannes Lercher**

Pacific Northwest National Lab  
Technical University Munich

Mechanisms and dynamics  
Controlled catalyst synthesis  
Heavy fossil energy feedstocks  
Biologically derived feedstocks  
Conversion of CO<sub>2</sub> and H<sub>2</sub>O

**Panel 1:** Diversified Energy Feedstocks and Carriers  
**Panel 2:** Novel Approaches to Energy Transformations  
**Panel 3:** Advanced Chemical Conversion Approaches  
**Panel 4:** Cross-cutting Capabilities and Challenges in  
Synthesis, Characterization, Theory and Computation

# Plenary Presentations at May Workshop



Catalyst Design for Sustainable Production of Fuels and Chemicals

Jens Norskov (Stanford)

The Nexus of Reaction Mechanism and Dynamic Materials Properties in Designing Catalytic Processes

Cynthia Friend (Harvard)



Opportunities for Catalysis in Utilization of Biomass Resources

Jim Dumesic (Wisconsin)

Creating New Economic Advantages from US Oil and Gas

Jim Rekoske (UOP)

Lessons From the Quest for Cellulosic Biofuels.....

Kim Johnson (Shell)



Impact of Catalytic Technology on Use of Renewable Energy Resources

Reuben Sarkar (EERE)

Frontiers, Challenges and Opportunities in Biological and Bio-Inspired Catalysis

Russ Hille (UC Riverside)



# Panel Breakout Sessions at May Workshop



## Panel 1: *Diversified Energy Feedstocks and Carriers*

**Geoffrey Coates**, Cornell University

**Enrique Iglesia**, University of California Berkeley, and Lawrence Berkeley National Laboratory



## Panel 2: *Novel Approaches to Energy Transformations*

**Morris Bullock**, Pacific Northwest National Laboratory

**Thomas Jaramillo**, Stanford University, and Stanford Linear Accelerator Center



## Panel 3: *Advanced Chemical Conversion Approaches*

**Maria Flytzani-Stephanopoulos**, Tufts University

**Cathy Tway**, Dow Chemical Company

**Daniel Resasco**, University of Oklahoma



## Panel 4: *Crosscutting Capabilities and Challenges: Synthesis, Theory, and Characterization*

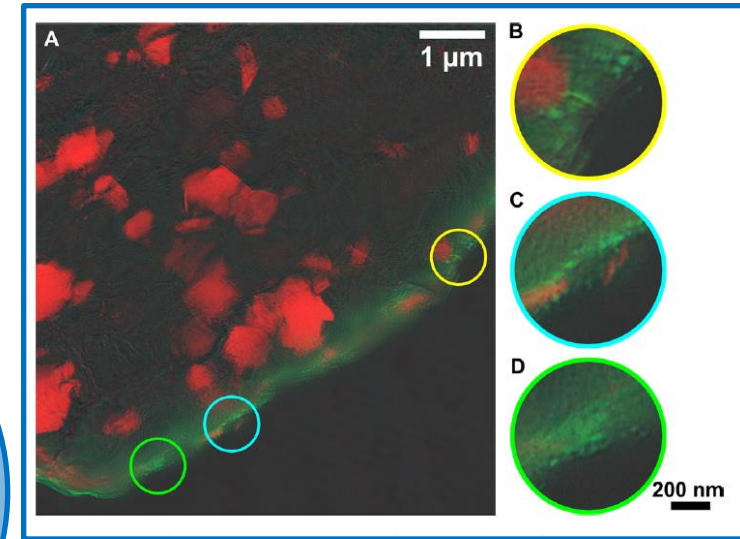
**Karena Chapman**, Argonne National Laboratory

**Victor Batista**, Yale University

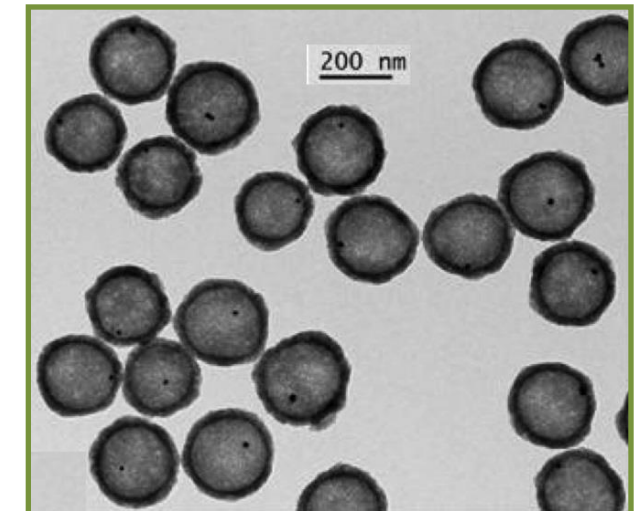
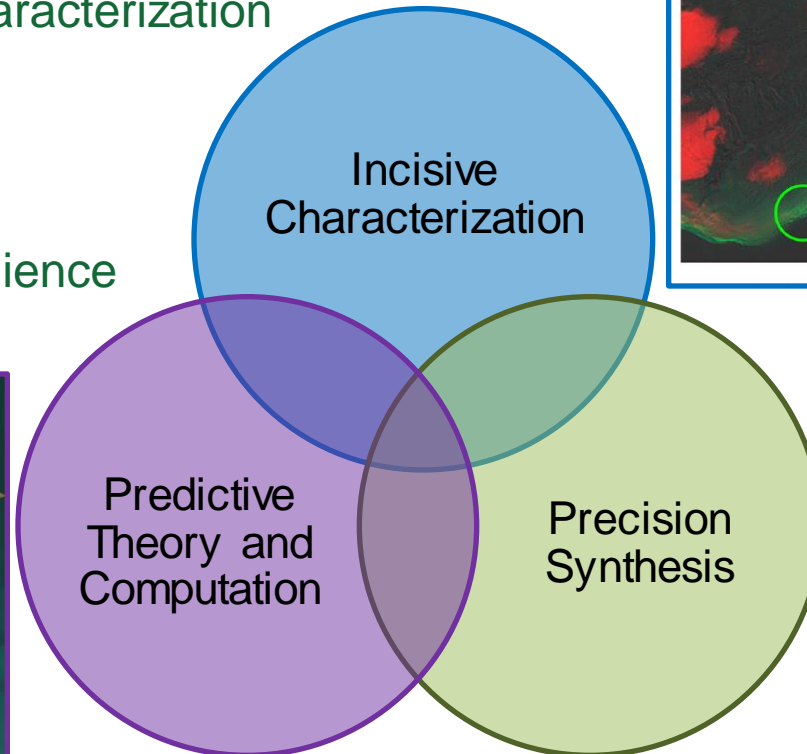
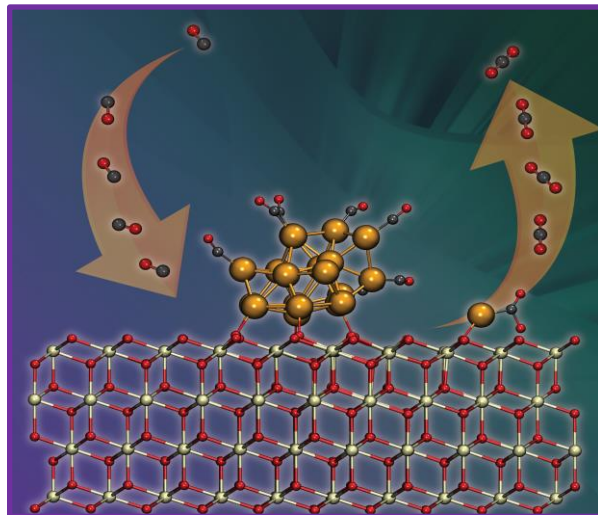
**Sheng Dai**, Oak Ridge National Laboratory

# Transformative Capabilities for Catalysis: Synthesis, Computation, and Characterization

- Identify rare and/or metastable active states
- Span multiple time/length-scales
- Conduct multimodal *operando* characterization
- Resolve reaction mechanisms
- Control interfacial interactions
- Incorporate cooperativity and resilience



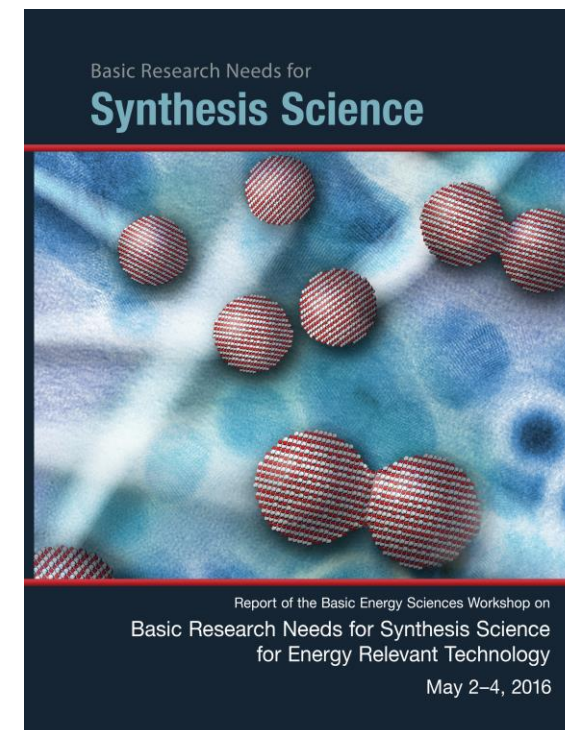
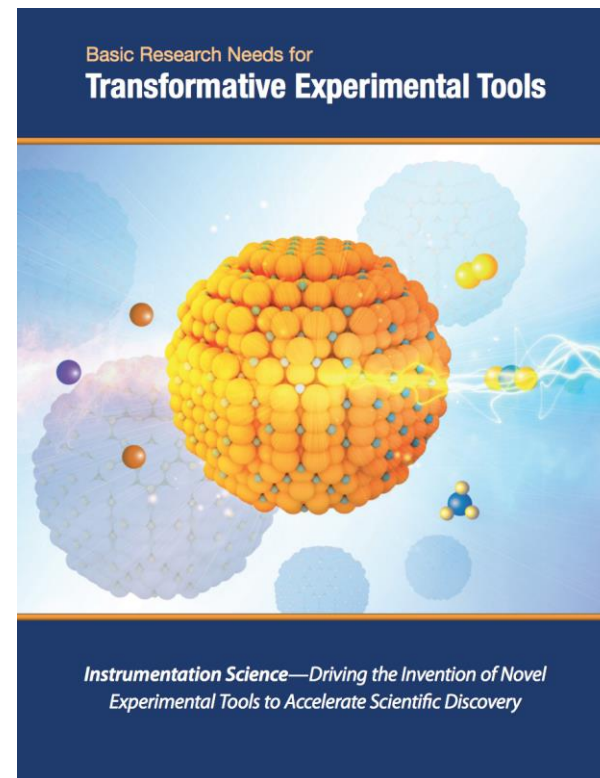
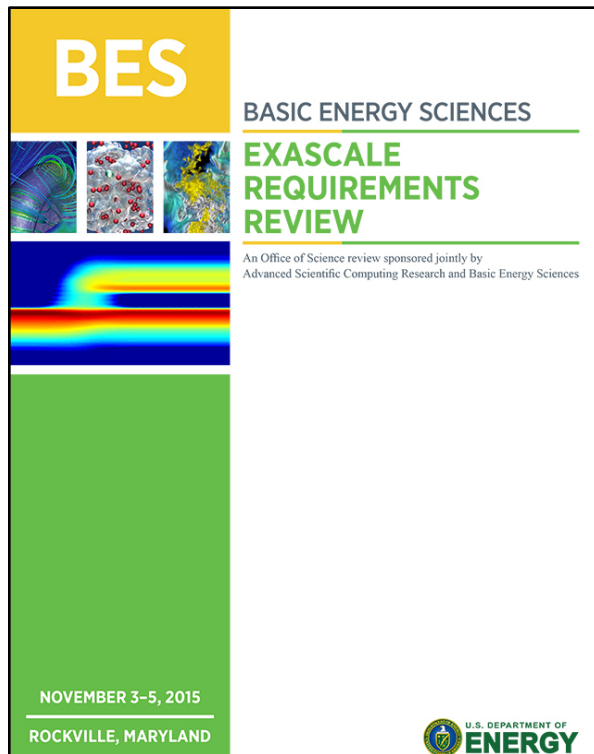
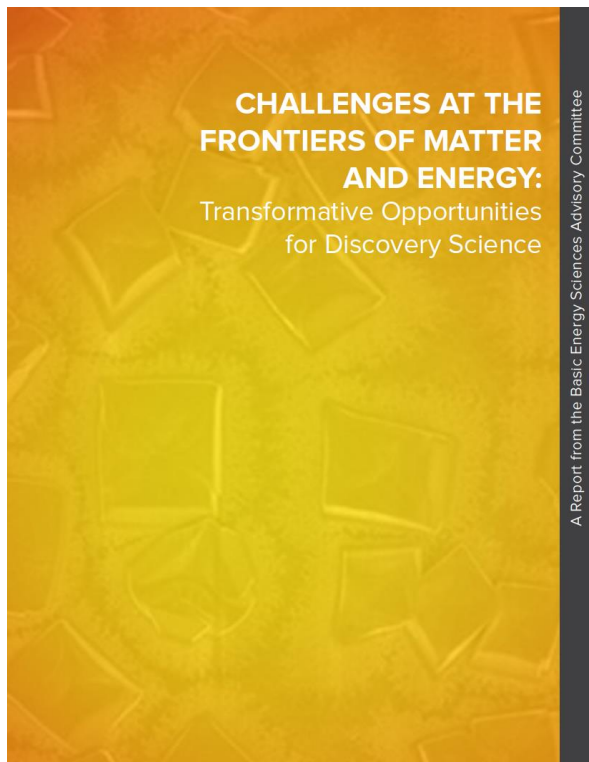
Meier, Weckhuysen et al., *ACS Catal.*, **2016**, 6, 2178.



Zaera et al., *Surf. Sci.* 2016, 648, 150



# Recent DOE Reports Emphasizing Synthesis, Computation, and Characterization

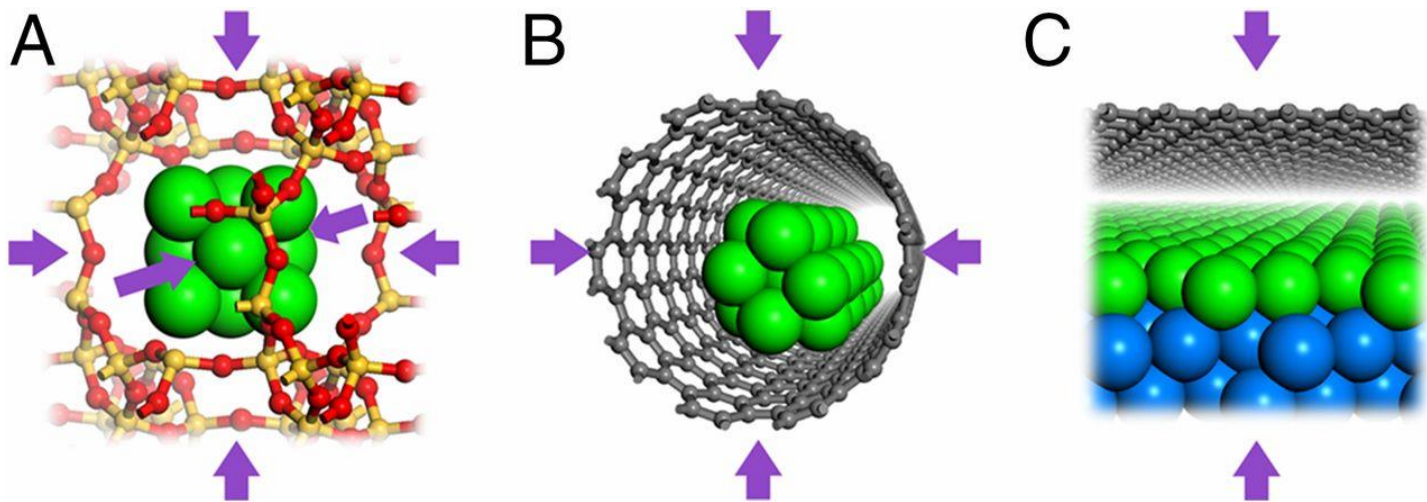


# Priority Research Direction 1

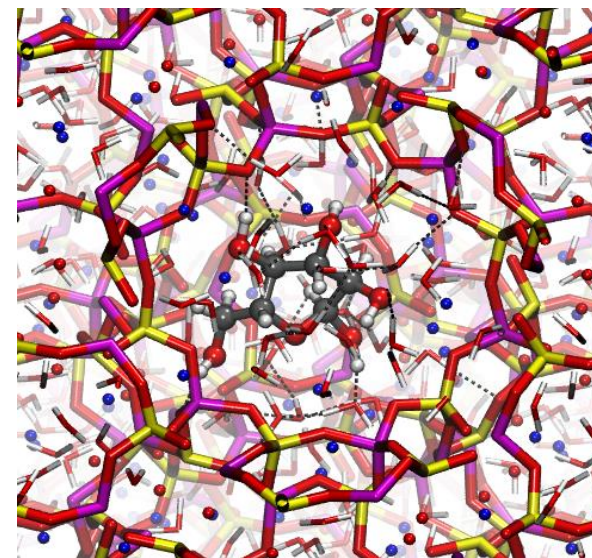
## Construct catalyst architectures to incorporate strong and weak interactions that organize matter and space beyond the binding site

*The extended structure of the active site includes not only the locations where the molecular orbitals of the reactants interact directly with the catalyst, but also nearby regions in space where the atomic organization in 3-dimensions serves to assemble and precisely position both reacting molecules and non-reacting components such as solvent molecules and counter-ions to achieve selective chemical transformations at high rates.*

Metal nanoparticle catalyst confinement, (A) inside a zeolite pore; (B) inside a carbon nanotube; (C) beneath a graphene sheet



Glucose solvation by water in a zeolite supercage

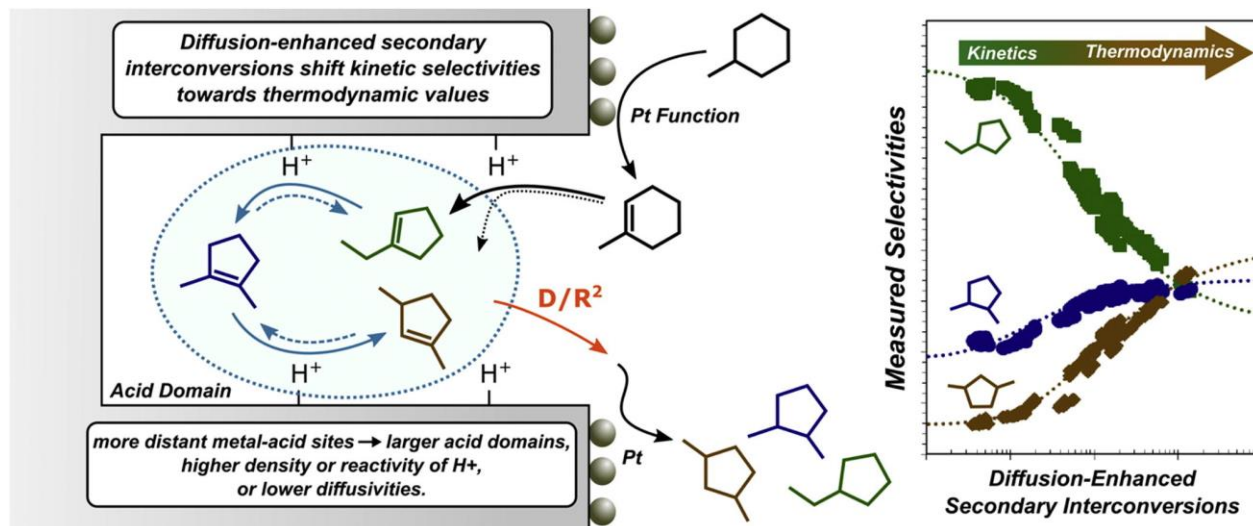




# Constructing Cooperative Catalyst Architectures

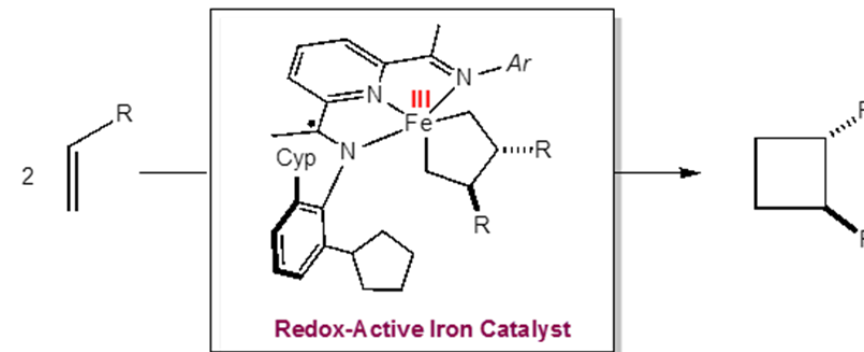
- How do we map the cooperative roles in catalyst-substrate ensembles at an atomistic level, using powerful emerging characterization and theory methods?
- How do we design synthesis methods that direct the organization of structural and functional motifs in the extended catalyst architecture so as to influence the ground and transition states of reacting molecules?

Effect of metal-acid site proximity on activity and selectivity in bifunctional catalysts



Iglesia et al., *J. Catal.* **2016**, 344, 817.

Redox-active ligands enable thermally-forbidden cyclodimerization



Chirik et al., *Science* **2015**, 349, 960.

Bimetallic cooperativity in polyol dehydration

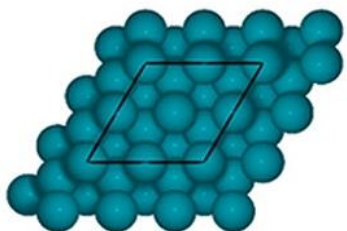


Davis et al., *ACS Catal.* **2015**, 5, 5679

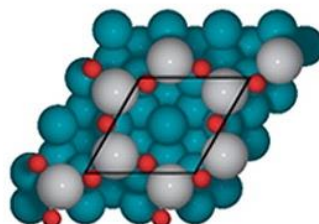
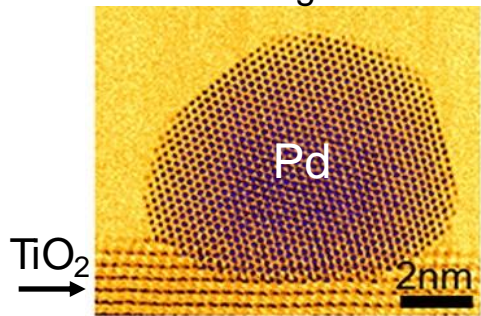
## Priority Research Direction 2

# Control the dynamic evolution of catalysts by influencing the rates and directing the pathways for reorganization

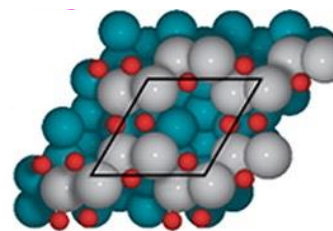
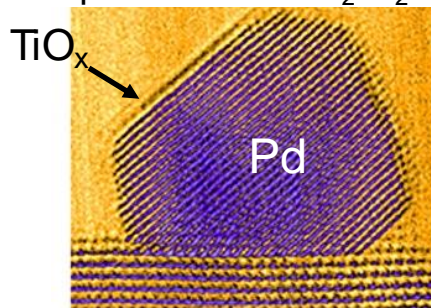
*Catalysts are inherently dynamic materials whose local and extended structures change continuously, beginning when the components are assembled into a catalytically active architecture and continuing as materials interact with reaction mixtures.*



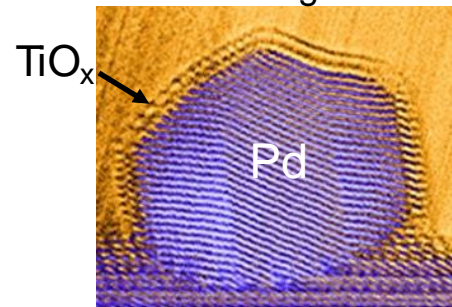
No  $\text{TiO}_x$  layer on Pd nanoparticle under oxidizing conditions



$\text{TiO}_x$  monolayer on Pd nanoparticle under  $\text{H}_2/\text{O}_2$  mixture



$\text{TiO}_x$  bilayer on Pd nanoparticle under reducing conditions

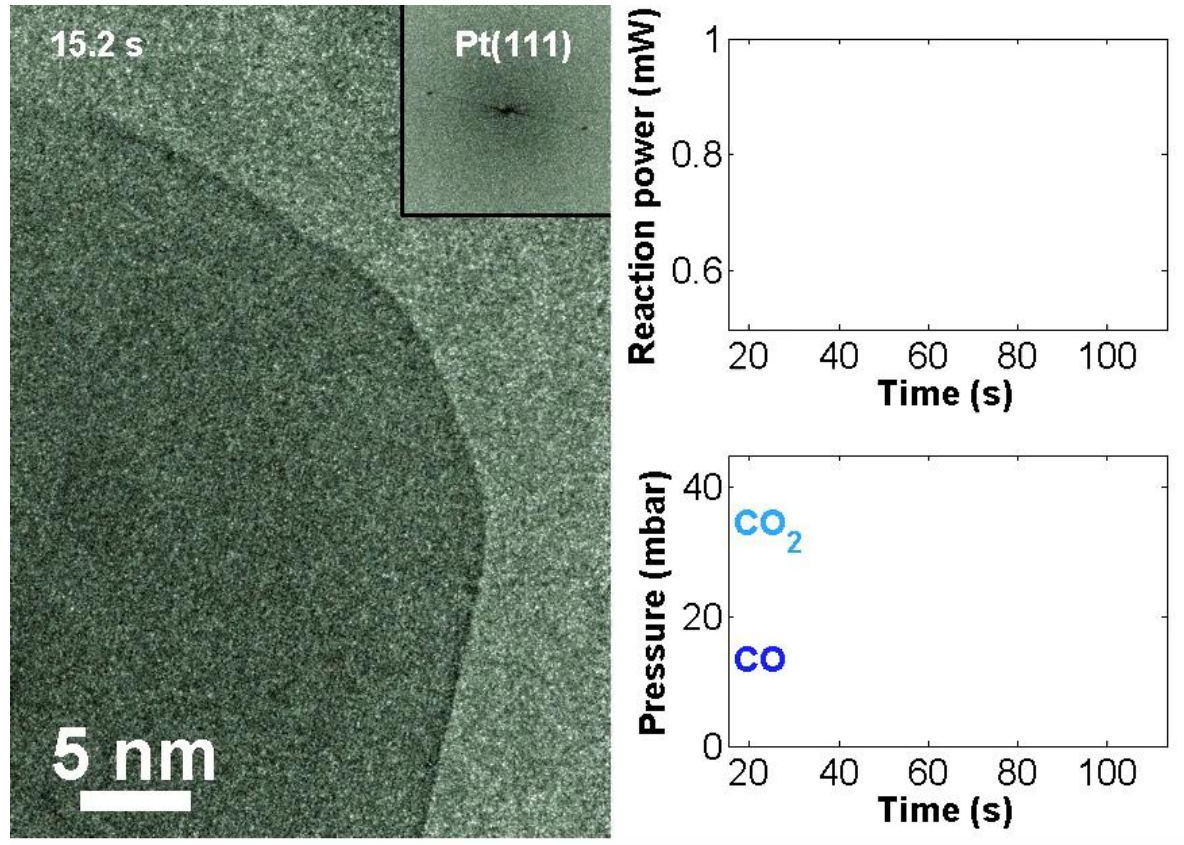




# Observing and Directing Complex Catalyst Behavior

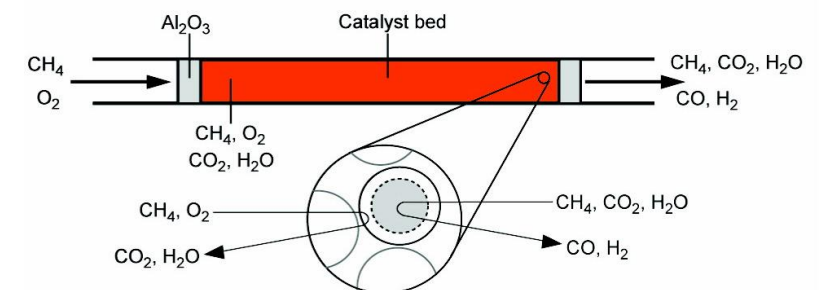
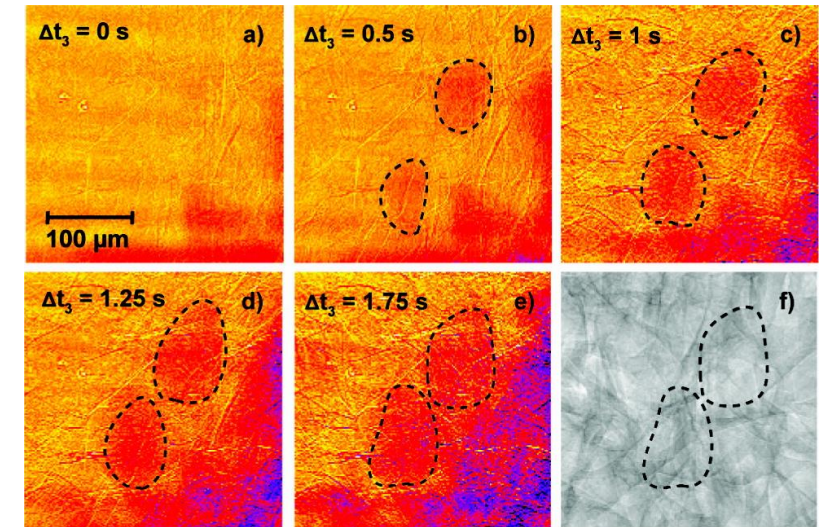
- How do we monitor changes in catalysts in real time and relate their consequences to reactivity and selectivity?
- How do we design catalysts to control their rates of activation, deactivation, and reactivation?

Time-resolved TEM images of Pt nanoparticle exposed to CO/O<sub>2</sub> at 727 K



Kooyman, Helveg, et al. *Nature Mater.*, 2014, 13, 884-890.

Reduction/ignition of individual Pt-Rh nanoparticles during catalytic partial oxidation of methane



Grunwaldt et al. *J. Phys. Chem. C*, 2009, 113, 3037.

## Priority Research Direction 3

# Decode complex networks of reactions, and integrate catalytic reactions with molecular transport and separations

*Emerging chemical fuels and feedstocks are often mixtures that present challenges for catalysis in terms of their inherent complexity and variability. In addition, their highly distributed nature will require entirely new approaches to catalytic processing.*



Agriculture Residues



Flared Gas



Municipal Solid Waste



Pulp Waste



Wet Sludges



Fats, Oils, and Greases

Alternative feedstocks have huge energy potential, and pose unique challenges for catalyst technology.



Complex mixtures  
Distributed sources  
Solids handling

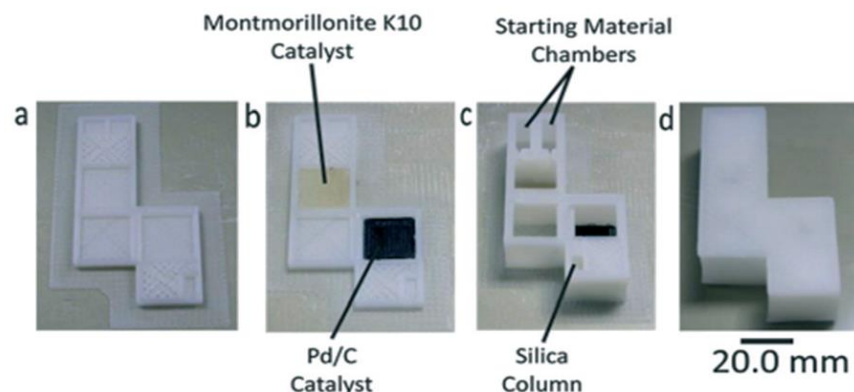
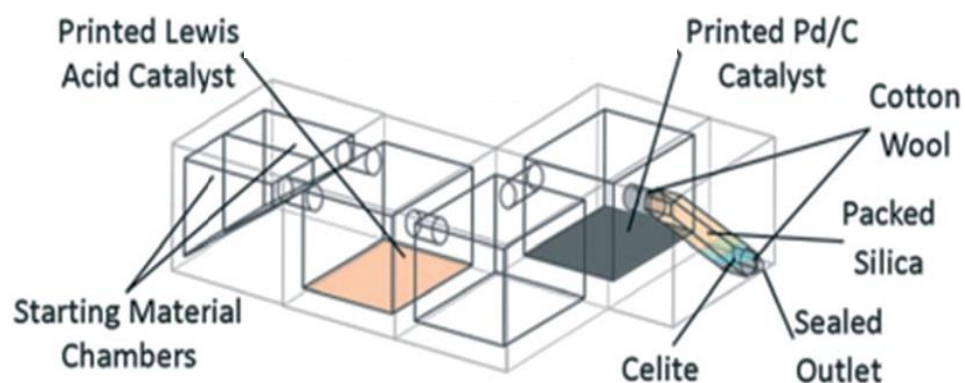




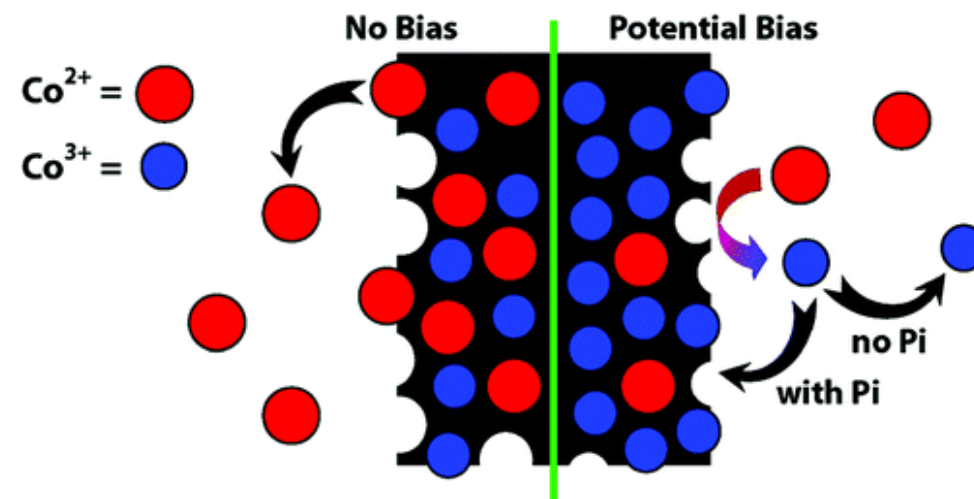
# Integration of Reaction with Separation and Design for Catalyst Resilience

- How do we design catalysts that adapt to transients in feed composition and reaction conditions, and that are capable of operating in highly distributed systems with minimal external control?

Achieving spatial control over catalysts, reactions and separations



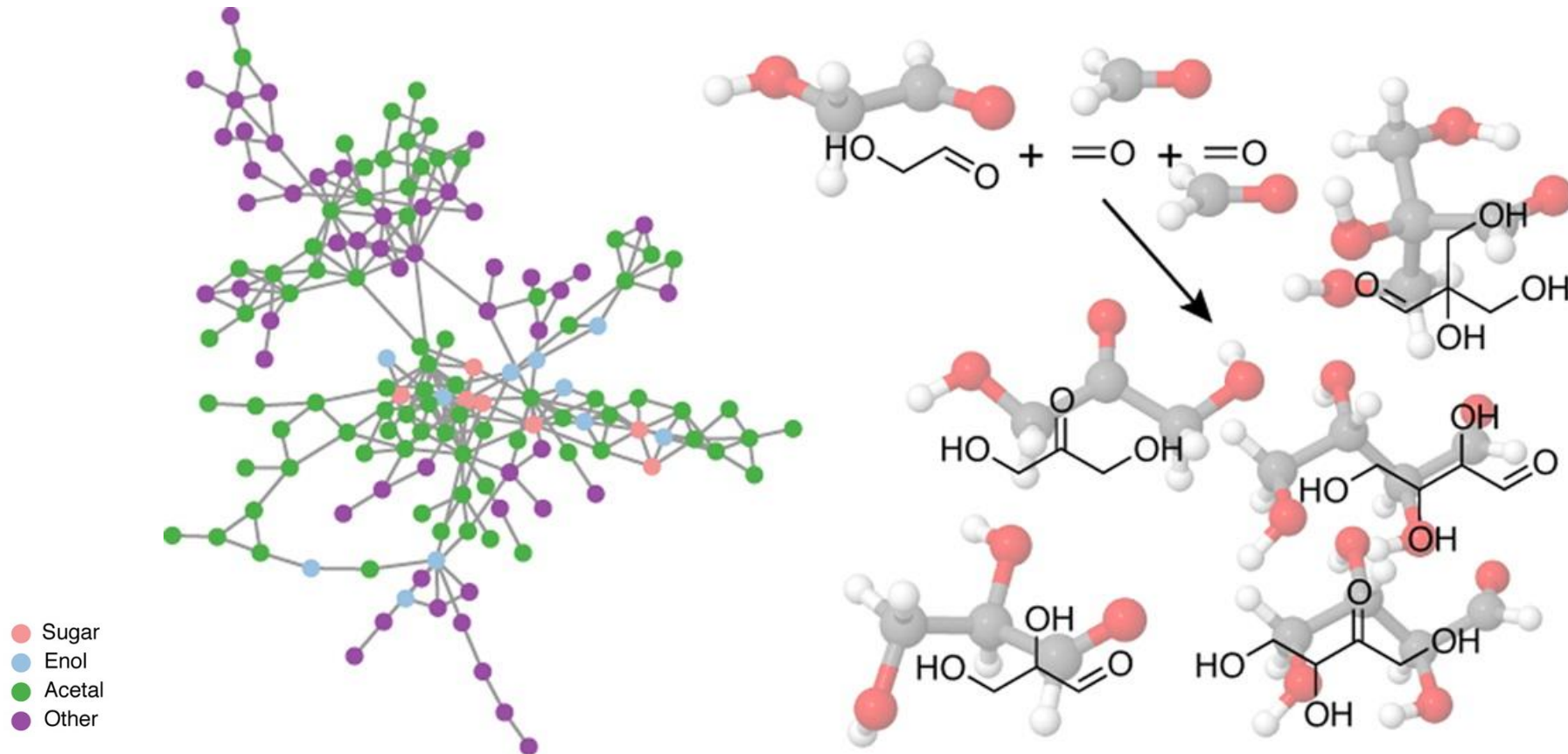
Self-healing oxygen-evolving catalyst



Nocera et al. *J. Am. Chem. Soc.* **2009**, 131, 3838.

# Combining Catalyst Discovery and Mechanisms with Systems-Level Insight

- How do we direct chemical transformations in multicomponent mixtures towards specific desired chemical products, via systems-level control of all the micro- and macro-kinetic factors that dictate rates in catalytic reaction cascades?



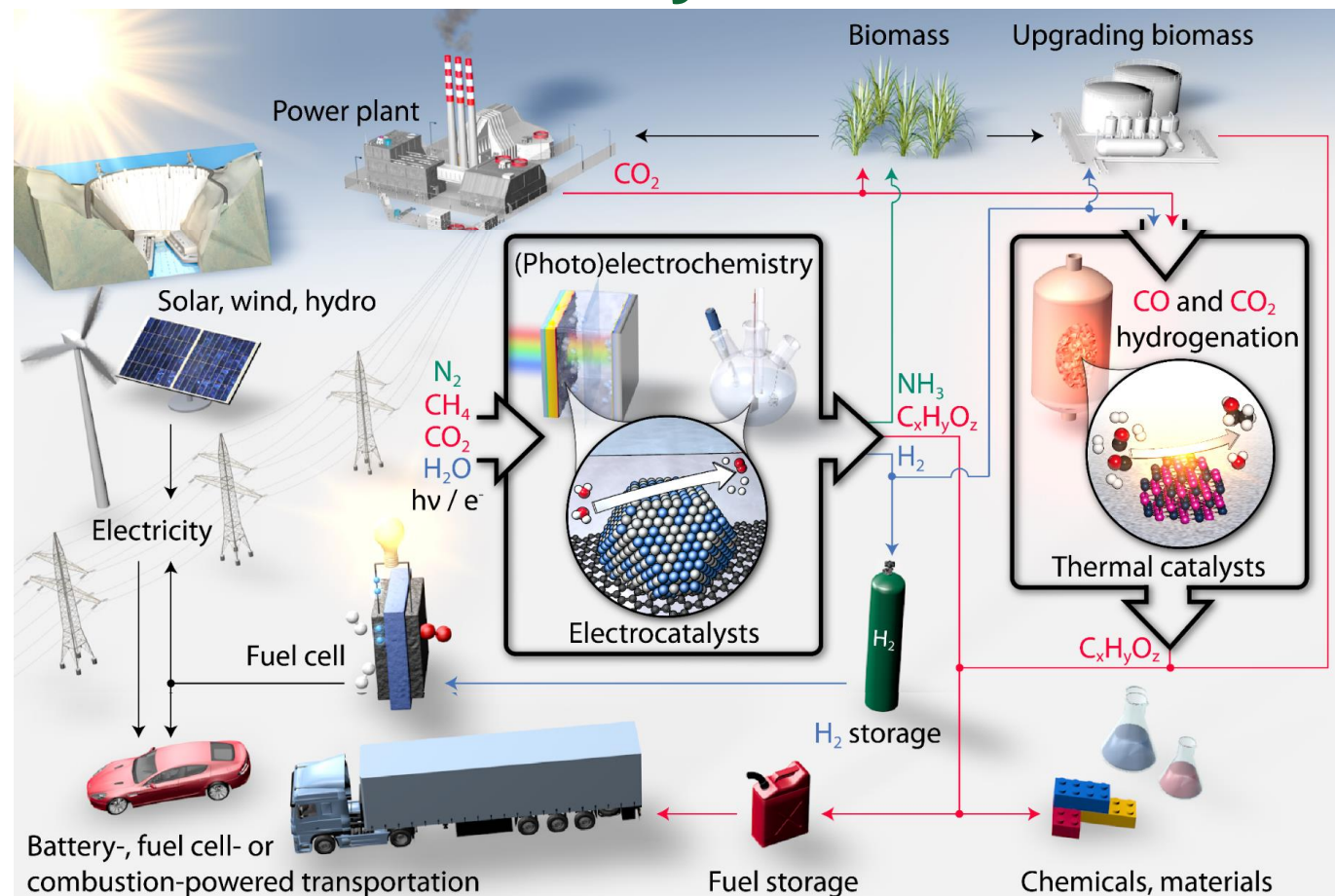
Construction of a reaction network for the *formose* reaction (self-condensation of formaldehyde in alkaline solution, resulting in a complex mixture of aldose and ketose sugars) using heuristic transformation rules.

## Priority Research Direction 4

# Design electrocatalyst systems to optimize electron-driven processes for precise chemical transformations efficiently under mild conditions

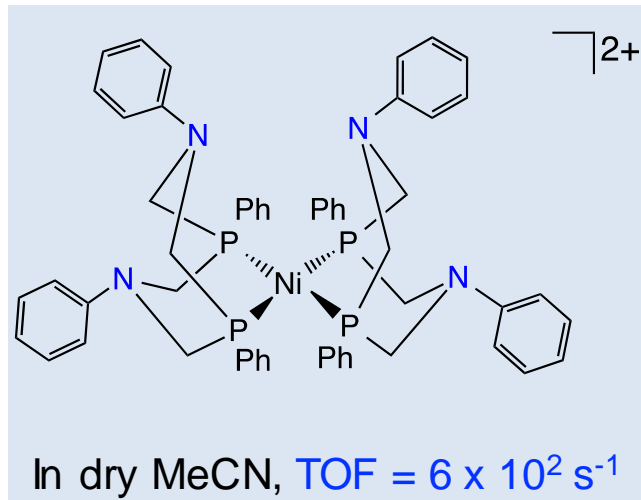
*The interconversion of chemical and electrical energy is based on the ability to store energy in chemical bonds and retrieve energy from those bonds by using the electron potential to control the directions and rates of chemical processes.*

*Electrochemical systems offer the possibility to drive chemical transformations under mild conditions, where thermal catalysis is inefficient.*

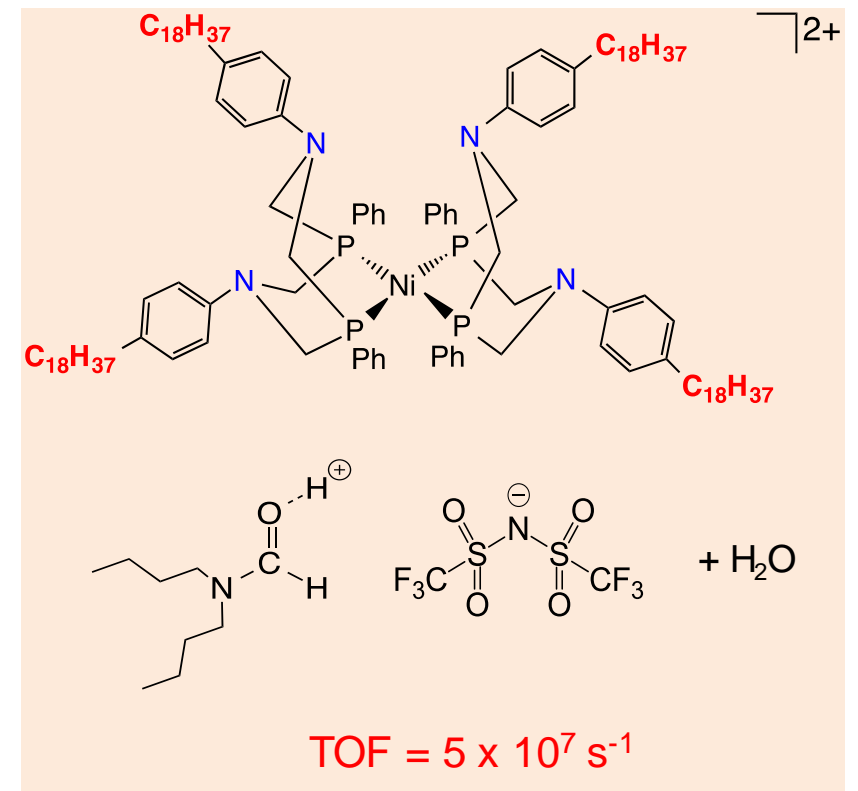


# Enhancing Activity in Molecular Electrocatalysts for H<sub>2</sub> Production

- How do we design conducting solid/solution interfaces, and coupled redox reaction chemistries, that catalyze electron-driven chemical processes with high precision and efficiency under mild conditions?
- How do we improve electrocatalysts for critical multi-electron, multi-proton reactions, and also expand the repertoire of chemical transformations that can occur effectively in electrochemical systems?
- How do we develop new electrocatalytic systems that circumvent known scaling relationships through detailed mechanistic understanding of catalytic pathways?



Detailed mechanistic understanding



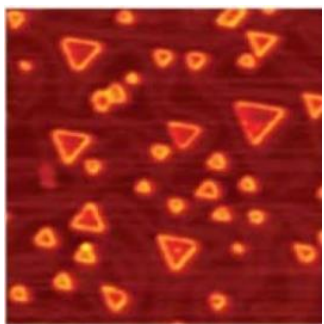
Roberts, DuBois et al., *J. Am. Chem. Soc.* **2011**, 133, 5861.

O'Hagan et al., *Angew. Chem. Int. Ed.* **2016**, 55, 13509.

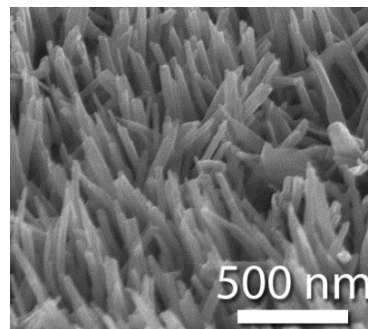


# Designing Nano-Structured Catalysts for H<sub>2</sub> Production

Increasing fraction of edge sites = higher activity

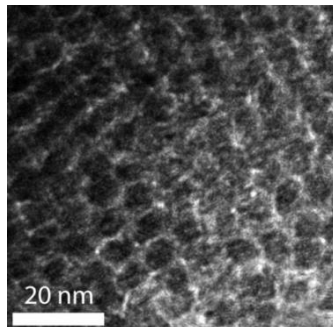


MoS<sub>2</sub> nanoparticles



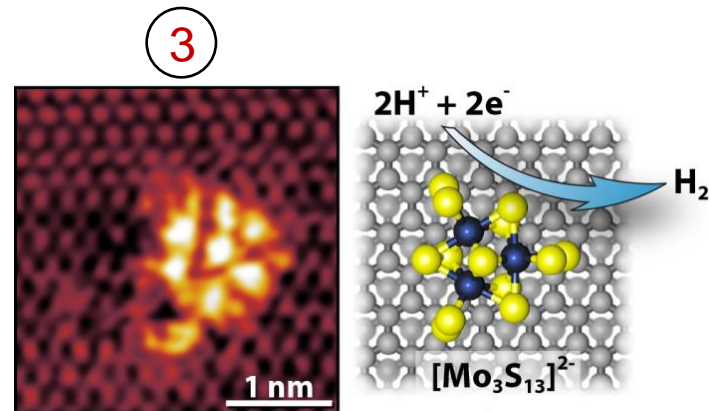
Core-shell MoO<sub>3</sub>-MoS<sub>2</sub> nanowires

1



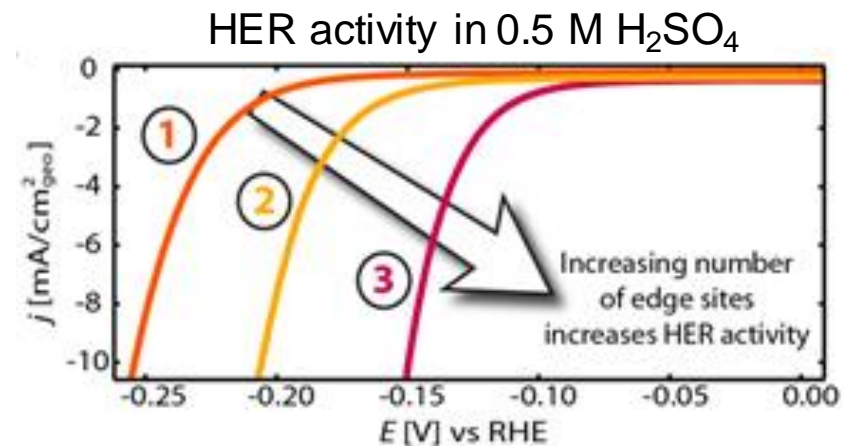
Mesoporous MoS<sub>2</sub> thin films

2



[Mo<sub>3</sub>S<sub>13</sub>]<sup>2-</sup> clusters

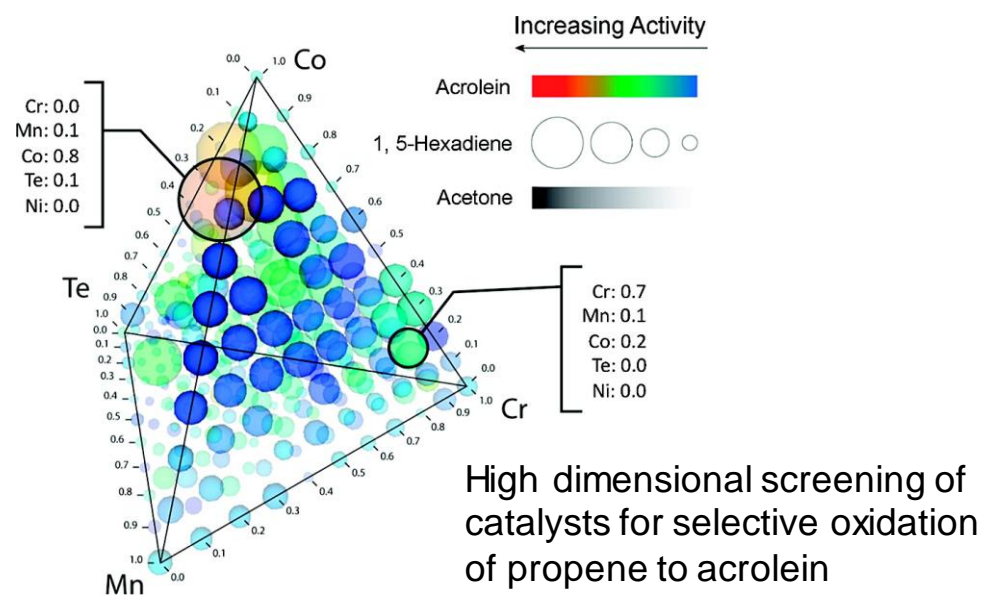
3



# Priority Research Direction 5

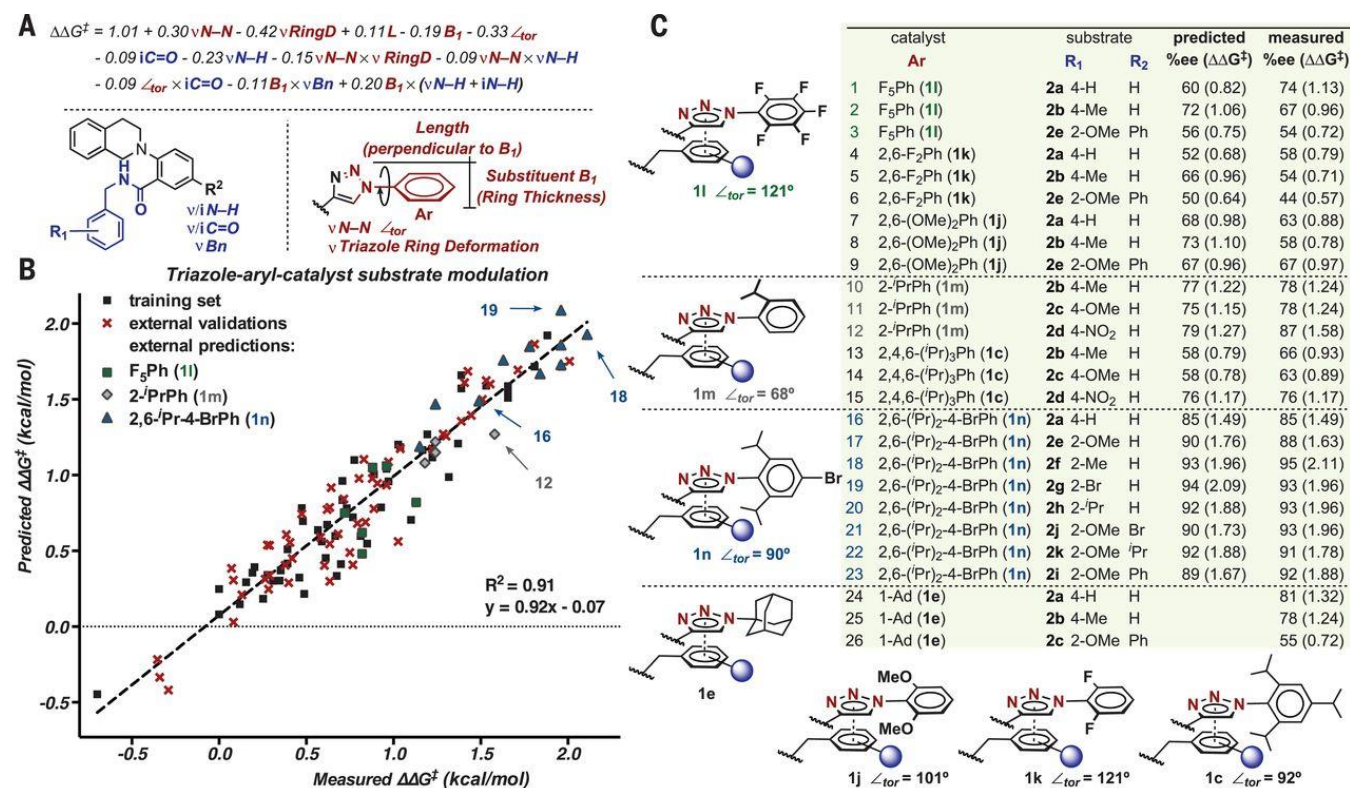
## Drive new catalyst discoveries using data science to resolve complex structure-function relations, predict new catalyst properties, and design more incisive experiments

The complex coupling of many variables that govern catalyst reactivity and its evolution in time makes it challenging to discern relationships between catalyst structure/composition and performance.



Rajan et al. *J. Comb. Chem.* **2009**, *11*, 385

### Multivariate analysis of catalytic C-N bond formation



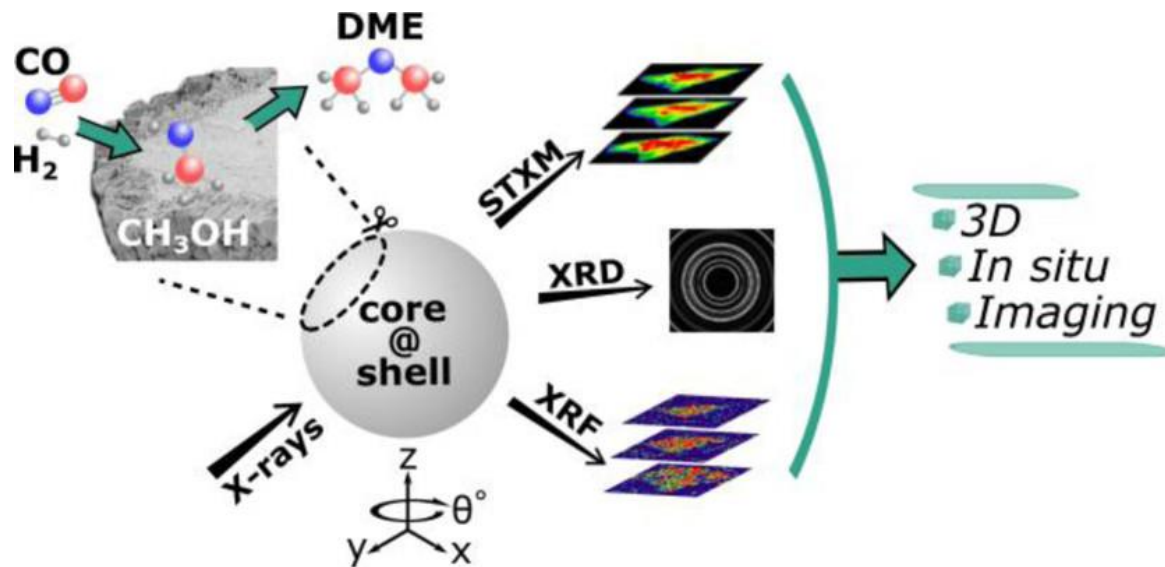
Milo, Neel et al. *Science* **2015**, *347*, 6223



# Mining Complex Information Using Data Science Approaches

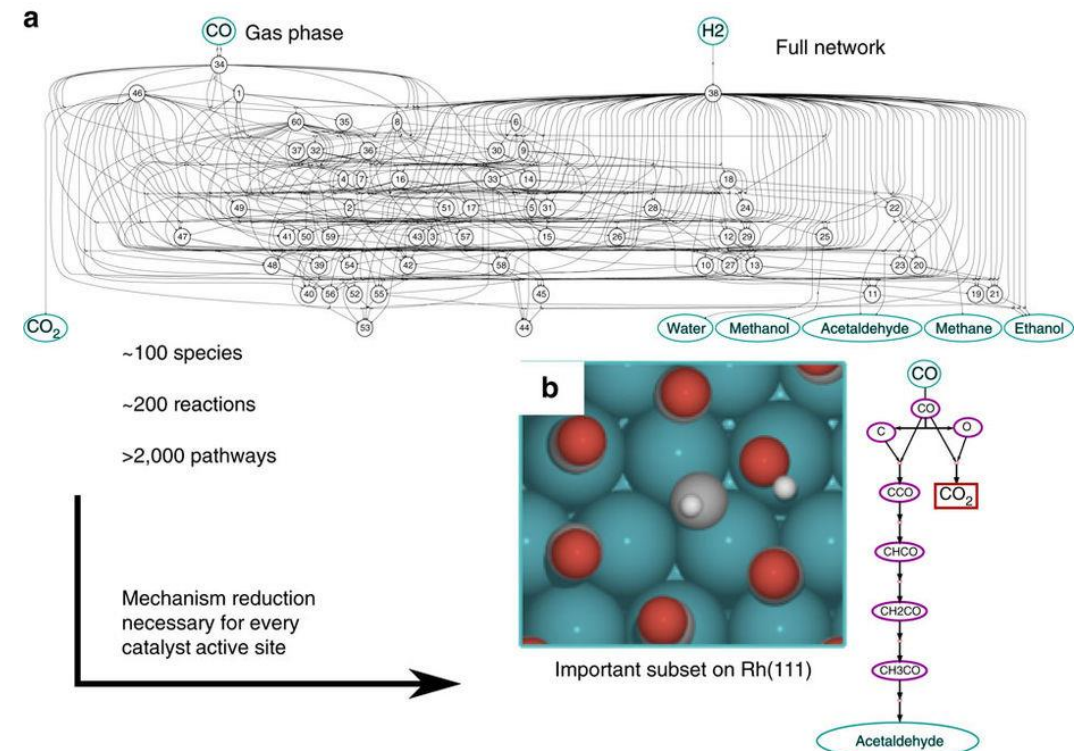
- How do we deploy the tools of machine learning to find and extract robust, often unexpected structure-activity relations from large, heterogeneous datasets?
- How do we use this information to predict effective combinations of catalytic structural components and reaction cascades, leading to entirely new catalyst formulations?

In situ multimodal 3D chemical imaging of a hierarchically structured core@shell catalyst for methanol synthesis



Grunwaldt et al. *J. Am. Chem. Soc.* **2017**, 139, 7855

Machine learning simplifies reaction network for syngas (CO+H<sub>2</sub>) conversion to acetaldehyde over Rh(111)

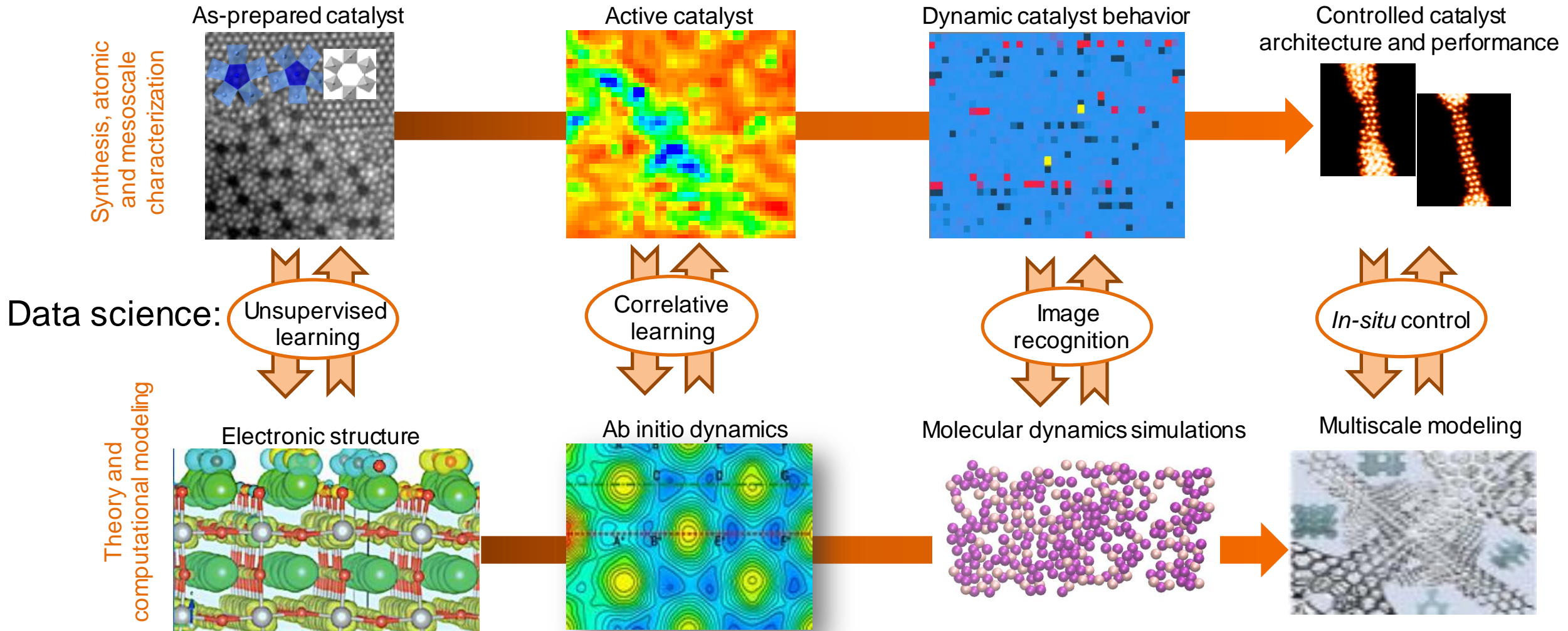


Bligaard, Nørskov, et al. *Nat. Commun.* **2017**, 8, 14621

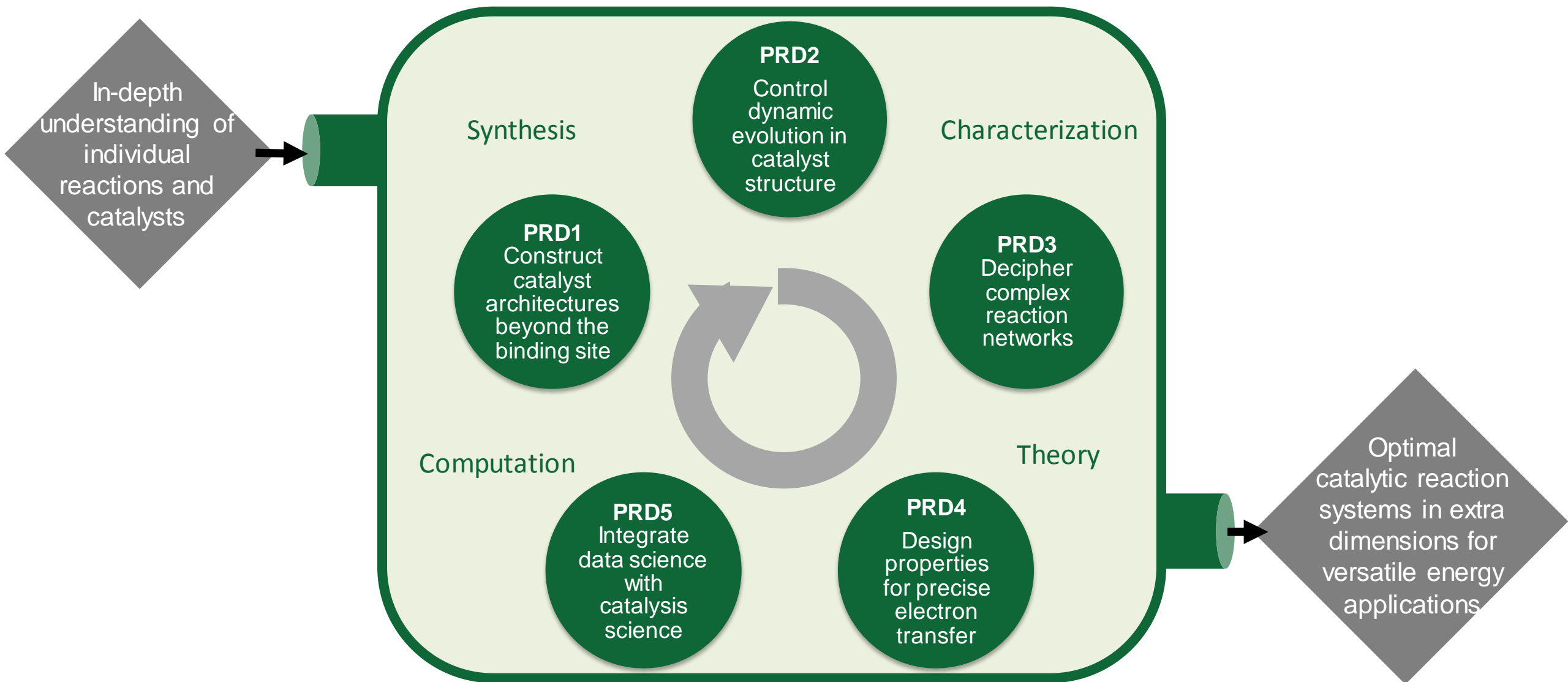


# Extracting More Detailed Information from Experiments

- How do we use data science tools to design better experiments with higher resolution and greater sensitivity, and extract more information from them?



# Embracing Complexity in Catalysis Science



# Progress towards Final Report

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## Appendix I: Technology Perspectives Resource Document

### Draft Panel Reports and PRDs on Sharepoint Site

Weekly videoconferences involving ca. 30 people involved in writing

Goal is to have first drafts of Brochure and Final Report by end of July